

Updates to photolysis calculations in WRF-Chem and evaluation with measurements

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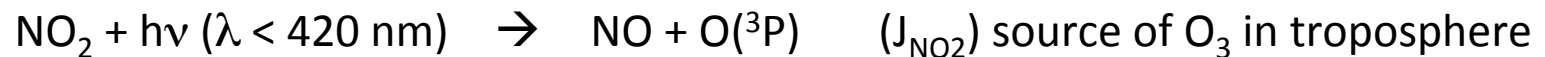
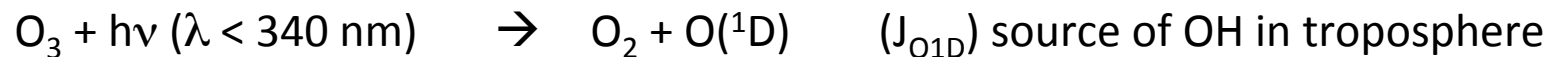


NCAR

Introduction

Photolysis: - chemical dissociation caused by absorption of solar radiation;
- an essential component of air quality models

Example :



Photolysis rates :

$$J (\text{s}^{-1}) = \int_{\lambda} F(\lambda) \sigma(\lambda) \phi(\lambda) d\lambda$$

$F(\lambda)$ = spectral actinic flux, quanta $\text{cm}^{-2} \text{s}^{-1} \text{nm}^{-1}$

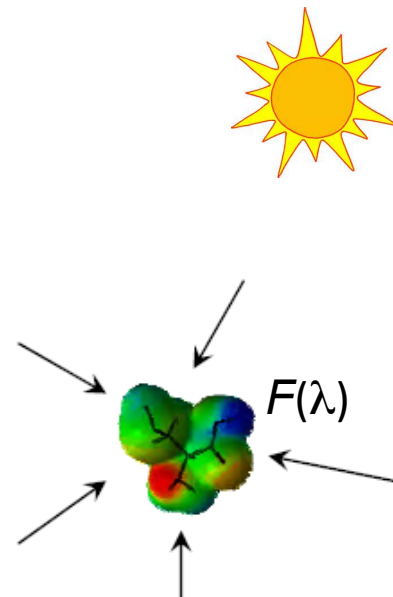
$\propto$ probability of photon near molecule.

$\sigma(\lambda)$ = absorption cross section, $\text{cm}^2 \text{molec}^{-1}$

$\propto$ probability that photon is absorbed.

$\phi(\lambda)$ = quantum yield, molec quanta^{-1}

$\propto$ probability that absorbed photon causes dissociation.



What do we need to know to predict photolysis reactions?

Compilations of Cross Sections & Quantum Yields

MPI-Mainz: <http://www.atmosphere.mpg.de/enid/2295>

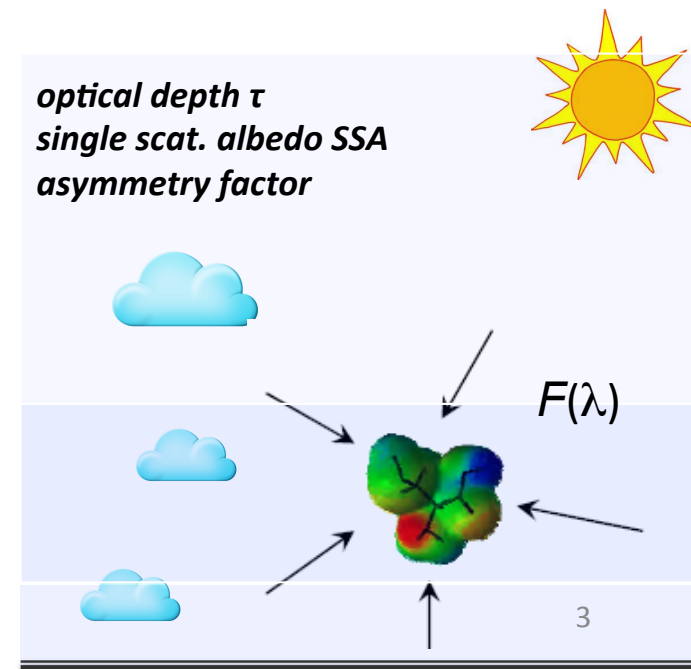
NASA JPL: <http://jpldataeval.jpl.nasa.gov/>

Actinic flux (F) is spherically integrated radiation ($\neq$ irradiances on the surface)

$$F = \int_0^\pi \int_0^{2\pi} I(\theta, \varphi) \sin \theta \, d\varphi \, d\theta \quad (\text{Watts m}^{-2} \text{ or quanta s}^{-1} \text{ cm}^{-2})$$

Attenuations/enhancements of photolysis by:

- Molecules
 - Rayleigh scattering
 - Absorbing O_2 , O_3 , NO_2 , SO_2
- Clouds (COD ~ 1 -500) scattering
- Aerosols (AOD < 5) absorbing and scattering



Photolysis calculations in WRF-Chem : Current & updated

Several radiative transfer packages:

- phot_opt = 1 : TUV (140 λ s, delta-Eddington)
- phot_opt = 2 : Fast-J (17 λ s, 8-str Feautrier)
- phot_opt = 3 : F-TUV (17 λ s, correction factor, delta-Eddington)
- ⇒ phot_opt = 4: updated TUV (140 λ s, delta-Eddington – next release)

Limitations of current schemes & improvements :

- Cross sections & quantum yields data
 - are hard-coded and not up to date
 - ⇒ data are read from an input file wrf_tuv_xsqa.nc
 - ⇒ data are based on the latest version of the TUV model (V5.2, Jan 2016)
- Limited number of photolysis reactions
 - hard to add new reactions
 - ⇒ 109 photolysis reactions relevant for tropospheric and stratospheric chemistry (e.g. halogens chemistry)

List of available photolysis reactions in the updated TUV

*in moztart_mosaic_4bin

1	$O_2 \rightarrow O + O$	(J_o2)	31	$2-C_4H_9ONO_2 \rightarrow 2-C_4H_9O + NO_2$	
2	$O_3 \rightarrow O_2 + O(1D)$	(J_o1d)	32	$CH_3CHONO_2CH_3 \rightarrow CH_3CHOCH_3 + NO_2$	
3	$O_3 \rightarrow O_2 + O(3P)$	(J_o3p)	33	$CH_2(OH)CH_2(ONO_2) \rightarrow CH_2(OH)CH_2(O.) + NO_2$	
4	$HO_2 \rightarrow OH + O$		34	$CH_3COCH_2(ONO_2) \rightarrow CH_3COCH_2(O.) + NO_2$	
5	$H_2O_2 \rightarrow 2 OH$	(J_h2o2)	35	$C(CH_3)_3(ONO_2) \rightarrow C(CH_3)_3(O.) + NO_2$	
6	$NO_2 \rightarrow NO + O(3P)$	(J_no2)	36	$C(CH_3)_3(ONO) \rightarrow C(CH_3)_3(O) + NO$	
7	$NO_3 \rightarrow NO + O_2$		37	$CH_3CO(OONO_2) \rightarrow CH_3CO(OO) + NO_2$	(J_pan_a)
8	$NO_3 \rightarrow NO_2 + O(3P)$		38	$CH_3CO(OONO_2) \rightarrow CH_3CO(O) + NO_3$	(J_pan_b)
9	$N_2O \rightarrow N_2 + O(1D)$	(J_n2o)	39	$CH_3CH_2CO(OONO_2) \rightarrow CH_3CH_2CO(OO) + NO_2$	
10	$N_2O_5 \rightarrow NO_3 + NO + O(3P)$		40	$CH_3CH_2CO(OONO_2) \rightarrow CH_3CH_2CO(O) + NO_3$	
11	$N_2O_5 \rightarrow NO_3 + NO_2$	(J_n2o5b)	41	$CH_2=CHCHO \rightarrow Products$	
12	$HNO_2 \rightarrow OH + NO$		42	$CH_2=C(CH_3)CHO \rightarrow Products$	(J_macr)
13	$HNO_3 \rightarrow OH + NO_2$	(J_hno3)	43	$CH_3COCH=CH_2 \rightarrow Products$	(J_mvkc)
14	$HNO_4 \rightarrow HO_2 + NO_2$	(J_hno4)	44	$HOCH_2CHO \rightarrow CH_2OH + HCO$	(J_glyald_a)
15	$NO_3-(aq) \rightarrow NO_2(aq) + O^-$		45	$HOCH_2CHO \rightarrow CH_3OH + CO$	(J_glyald_b)
16	$NO_3-(aq) \rightarrow NO_2-(aq) + O(3P)$		46	$HOCH_2CHO \rightarrow CH_2CHO + OH$	(J_glyald_c)
17	$CH_2O \rightarrow H + HCO$	(J_ch2or)	47	$CH_3COCH_3 \rightarrow CH_3CO + CH_3$	(J_ch3coch3)
18	$CH_2O \rightarrow H_2 + CO$	(J_ch2om)	48	$CH_3COCH_2CH_3 \rightarrow CH_3CO + CH_2CH_3$	(J_mek)
19	$CH_3CHO \rightarrow CH_3 + HCO$	(J_ch3cho_a)	49	$CH_2(OH)COCH_3 \rightarrow CH_3CO + CH_2(OH)$	(J_hyac_a)
20	$CH_3CHO \rightarrow CH_4 + CO$	(J_ch3cho_b)	50	$CH_2(OH)COCH_3 \rightarrow CH_2(OH)CO + CH_3$	(J_hyac_b)
21	$CH_3CHO \rightarrow CH_3CO + H$	(J_ch3cho_c)	51	$CHOCHO \rightarrow HCO + HCO$	(J_gly_a)
22	$C_2H_5CHO \rightarrow C_2H_5 + HCO$		52	$CHOCHO \rightarrow H_2 + 2CO$	(J_gly_b)
23	$CH_3OOH \rightarrow CH_3O + OH$		53	$CHOCHO \rightarrow CH_2O + CO$	(J_gly_c)
24	$HOCH_2OOH \rightarrow HOCH_2O. + OH$	(J_pooh)	54	$CH_3COCHO \rightarrow CH_3CO + HCO$	(J_mgly)
25	$CH_3ONO_2 \rightarrow CH_3O + NO_2$		55	$CH_3COCOCH_3 \rightarrow Products$	
26	$CH_3(OONO_2) \rightarrow CH_3(OO) + NO_2$		56	$CH_3COOH \rightarrow CH_3 + COOH$	
27	$CH_3CH_2ONO_2 \rightarrow CH_3CH_2O + NO_2$		57	$CH_3CO(OOH) \rightarrow Products$	
28	$C_2H_5ONO_2 \rightarrow C_2H_5O + NO_2$		58	$CH_3COCO(OH) \rightarrow Products$	
29	$n-C_3H_7ONO_2 \rightarrow C_3H_7O + NO_2$		59	$(CH_3)_2NNO \rightarrow Products$	
30	$1-C_4H_9ONO_2 \rightarrow 1-C_4H_9O + NO_2$		60	$CF_2O \rightarrow Products$	

List of available photolysis reactions in the updated TUV

61	$\text{Cl}_2 \rightarrow \text{Cl} + \text{Cl}$	91	$\text{CF}_3\text{CF}_2\text{CHCl}_2 \text{ (HCFC-225ca)} \rightarrow \text{Products}$
62	$\text{ClO} \rightarrow \text{Cl} + \text{O(1D)}$	92	$\text{CF}_2\text{ClCF}_2\text{CHFCl} \text{ (HCFC-225cb)} \rightarrow \text{Products}$
63	$\text{ClO} \rightarrow \text{Cl} + \text{O(3P)}$	93	$\text{Br}_2 \rightarrow \text{Br} + \text{Br}$
64	$\text{ClOO} \rightarrow \text{Products}$	94	$\text{BrO} \rightarrow \text{Br} + \text{O}$
65	$\text{OCIO} \rightarrow \text{Products}$	95	$\text{HOBr} \rightarrow \text{OH} + \text{Br}$
66	$\text{ClOOCl} \rightarrow \text{Cl} + \text{ClOO}$	96	$\text{BrNO} \rightarrow \text{Br} + \text{NO}$
67	$\text{HCl} \rightarrow \text{H} + \text{Cl}$	97	$\text{BrONO} \rightarrow \text{Br} + \text{NO}_2$
68	$\text{HOCl} \rightarrow \text{HO} + \text{Cl}$	98	$\text{BrONO} \rightarrow \text{BrO} + \text{NO}$
69	$\text{NOCl} \rightarrow \text{NO} + \text{Cl}$	99	$\text{BrNO}_2 \rightarrow \text{Br} + \text{NO}_2$
70	$\text{ClNO}_2 \rightarrow \text{Cl} + \text{NO}_2$	100	$\text{BrONO}_2 \rightarrow \text{BrO} + \text{NO}_2$
71	$\text{ClONO} \rightarrow \text{Cl} + \text{NO}_2$	101	$\text{BrONO}_2 \rightarrow \text{Br} + \text{NO}_3$
72	$\text{ClONO}_2 \rightarrow \text{Cl} + \text{NO}_3$	102	$\text{BrCl} \rightarrow \text{Br} + \text{Cl}$
73	$\text{ClONO}_2 \rightarrow \text{ClO} + \text{NO}_2$	103	$\text{CH}_3\text{Br} \rightarrow \text{Products}$
74	$\text{CCl}_4 \rightarrow \text{Products}$	104	$\text{CHBr}_3 \rightarrow \text{Products}$
75	$\text{CH}_3\text{OCl} \rightarrow \text{CH}_3\text{O} + \text{Cl}$	105	$\text{CF}_2\text{Br}_2 \text{ (Halon-1202)} \rightarrow \text{Products}$
76	$\text{CHCl}_3 \rightarrow \text{Products}$	106	$\text{CF}_2\text{BrCl} \text{ (Halon-1211)} \rightarrow \text{Products}$
77	$\text{CH}_3\text{Cl} \rightarrow \text{Products}$	107	$\text{CF}_3\text{Br} \text{ (Halon-1301)} \rightarrow \text{Products}$
78	$\text{CH}_3\text{CCl}_3 \rightarrow \text{Products}$	108	$\text{CF}_2\text{BrCF}_2\text{Br} \text{ (Halon-2402)} \rightarrow \text{Products}$
79	$\text{CCl}_2\text{O} \rightarrow \text{Products}$	109	perfluoro 1-iodopropane $\rightarrow$ products
80	$\text{CClFO} \rightarrow \text{Products}$		
81	$\text{CCl}_3\text{F} \text{ (CFC-11)} \rightarrow \text{Products}$		
82	$\text{CCl}_2\text{F}_2 \text{ (CFC-12)} \rightarrow \text{Products}$		
83	$\text{CF}_2\text{ClCFCl}_2 \text{ (CFC-113)} \rightarrow \text{Products}$		
84	$\text{CF}_2\text{ClCF}_2\text{Cl} \text{ (CFC-114)} \rightarrow \text{Products}$		
85	$\text{CF}_3\text{CF}_2\text{Cl} \text{ (CFC-115)} \rightarrow \text{Products}$		
86	$\text{CHClF}_2 \text{ (HCFC-22)} \rightarrow \text{Products}$		
87	$\text{CF}_3\text{CHCl}_2 \text{ (HCFC-123)} \rightarrow \text{Products}$		
88	$\text{CF}_3\text{CHFCl} \text{ (HCFC-124)} \rightarrow \text{Products}$		
89	$\text{CH}_3\text{CFCl}_2 \text{ (HCFC-141b)} \rightarrow \text{Products}$		
90	$\text{CH}_3\text{CF}_2\text{Cl} \text{ (HCFC-142b)} \rightarrow \text{Products}$		

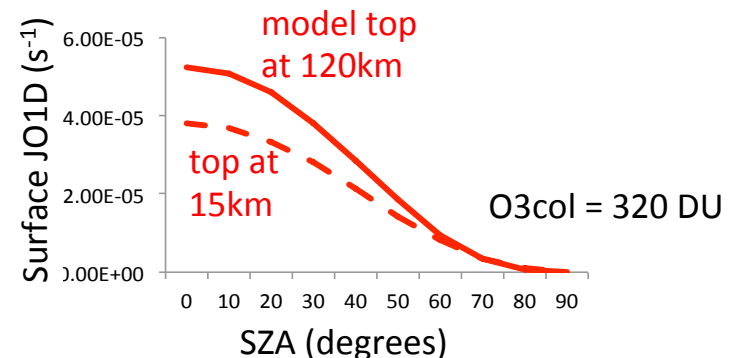
Additional file in KPP/mechanisms/\$mechanism/
\$mechanism.tuv.jmap

Correspondence j_wrfchem with available j_tuv

Photolysis calculations in WRF-Chem : Current & updated

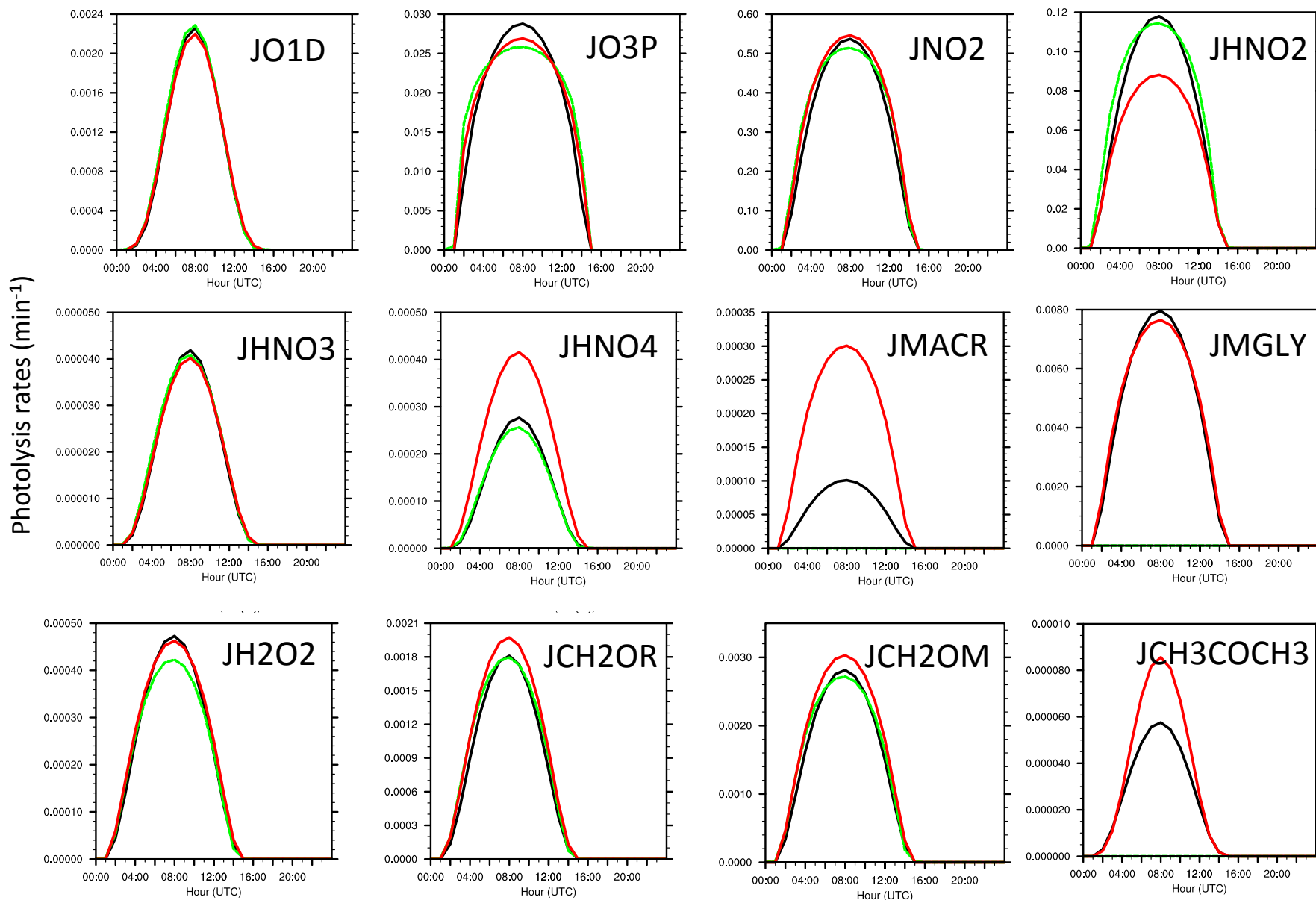
- Added routines in chem/
 - `module_phot_tuv.F`
 - `module_subs_tuv.F`
- Stratospheric ozone
 - f-TUV: climatology at the model top (input file `exo_coldens.nc`)
 - fast-J: specified value at the model top for the whole domain
 - TUV: specified value with extra levels above the model top

⇒ Updated TUV: uses ozone climatology distributed from model top to 50km (extra levels above the model top, standard atmosphere)
- Aerosols: calculated in the optics driver
 - Mixing rules for index of refraction;
 - Different core-shell options



Comparison of diurnal profiles in clear sky conditions (Boulder)

Old TUV (phot_opt=1); **New TUV (phot_opt=4)**; Fast-J (phot_opt=2)

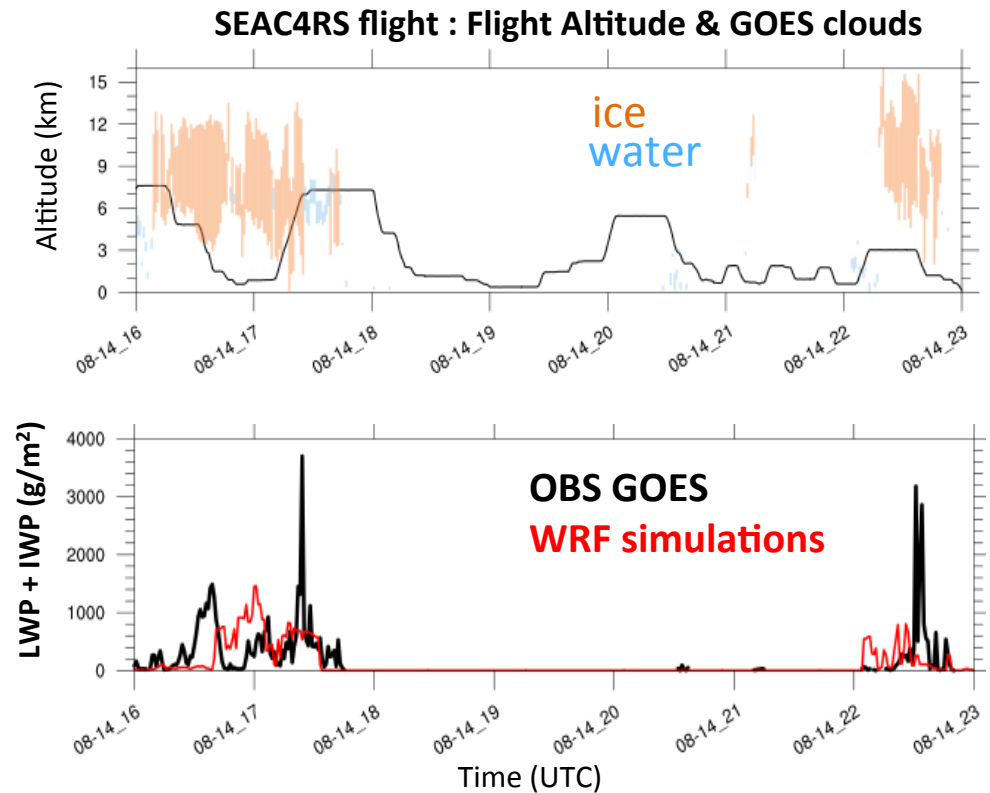
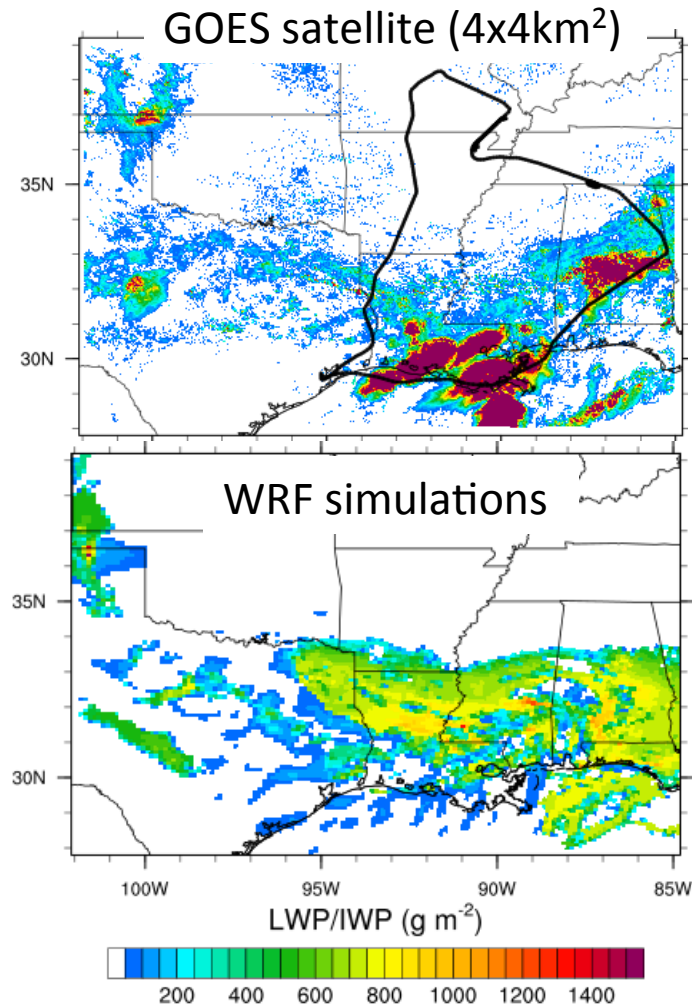


Photolysis calculations in WRF-Chem

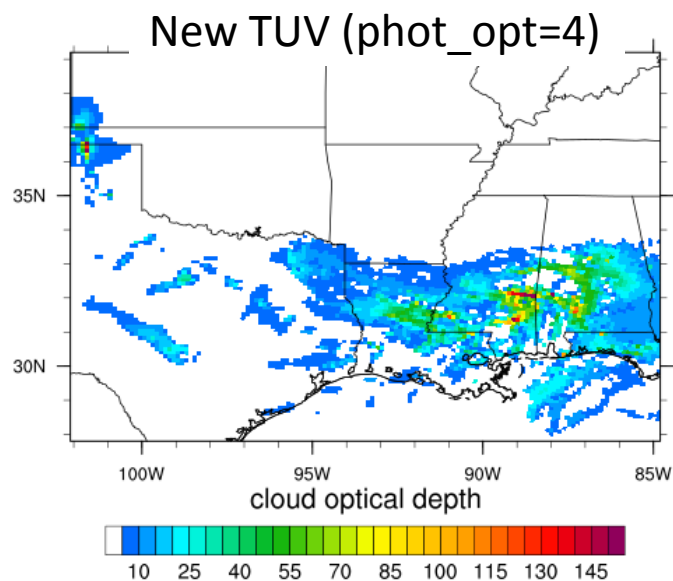
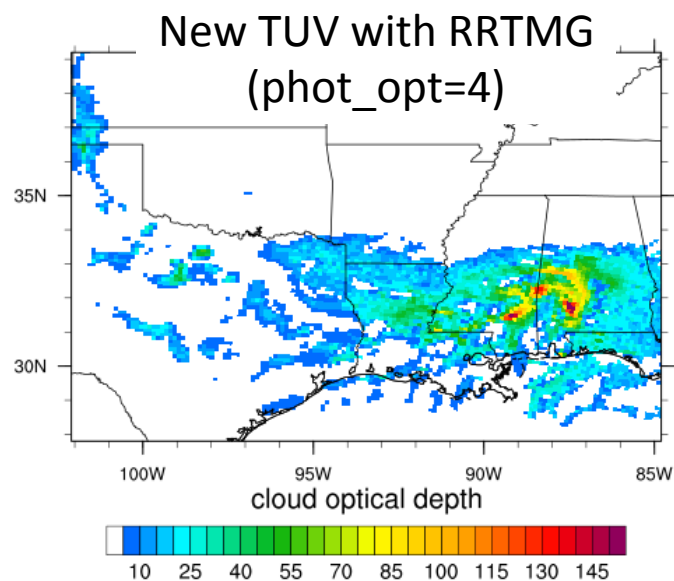
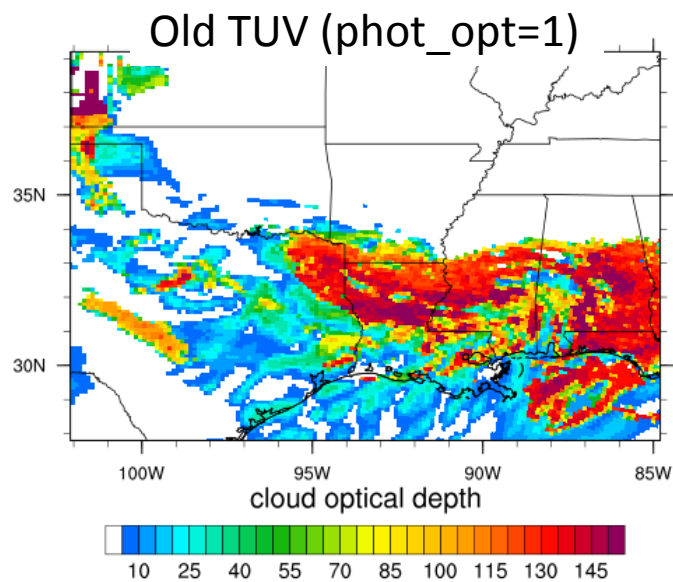
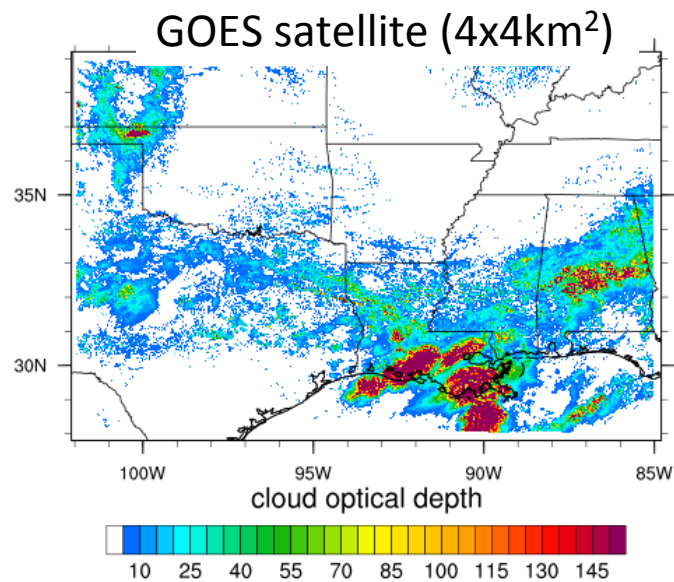
- Current treatment of clouds
 - Cloud optical properties are recalculated in photolysis scheme
 - Differ between schemes and from physics (e.g. RRTMG)
 - Sub-grid cloud overlap schemes
 - e.g. f-TUV (max overlap if vertically contiguous, random otherwise)
- Updated method 1 (phot_opt=4)
 - COD calculated from LWP/IWP and effective drop radius (Slingo 1989, similar to CAM radiation with $SSA = 0.9999$ and $f_{\text{assym}} = 0.85$)
 - Sub-grid cloud effects are represented as in Briegleb (1992)
($COD_{\text{subgrid}} = COD * FCLD^{3/2}$, equivalent to random overlap)
- Updated method 2 (phot_opt=4) uses RRTMG cloud vertical distribution
 - RRTMG adopts maximum-random overlap to distribute LWP/IWP randomly over wavelength bins based on cloud fraction.
 - Take COD that is averaged over the RRTMG wavelength bins.

Comparison with SEAC4RS flight of 14 Aug. 2013

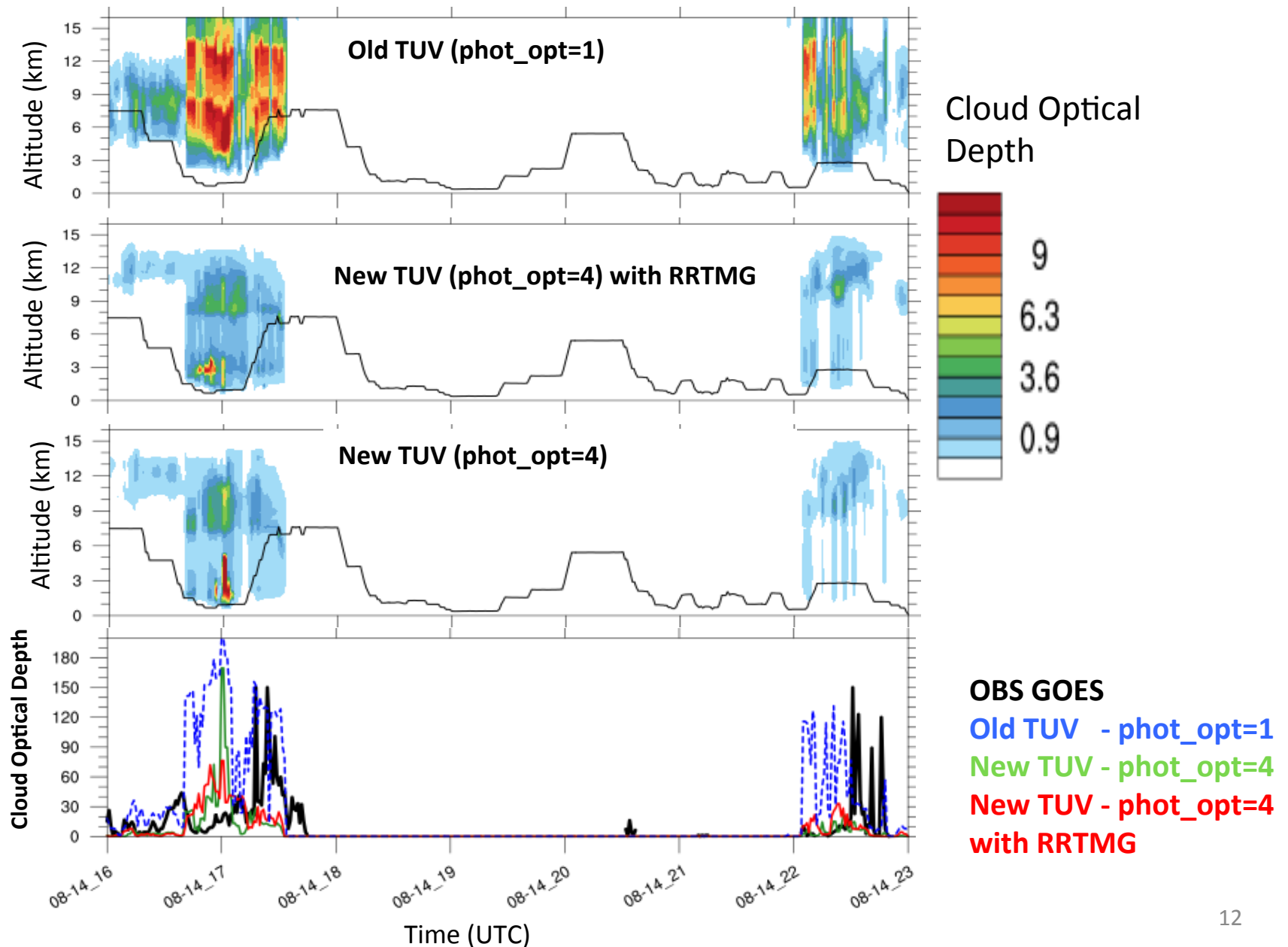
- WRF-Chem simulations : CONUS (12x12km²) during summer 2013
- MOZART-MOSAIC chemistry
- Morrison microphysics + CU Grell



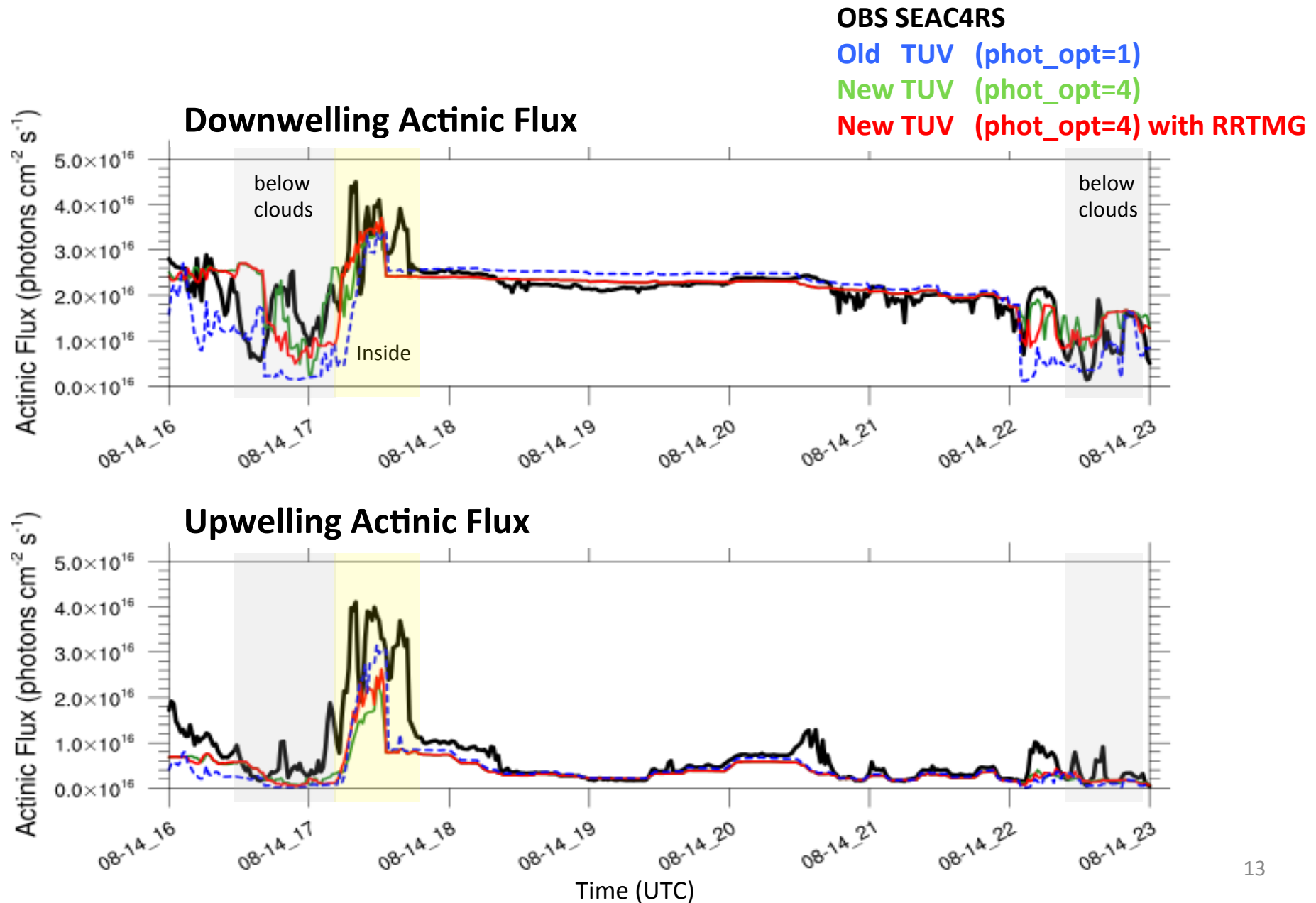
Effective cloud optical depths – 14 Aug. 2013 (18UTC)



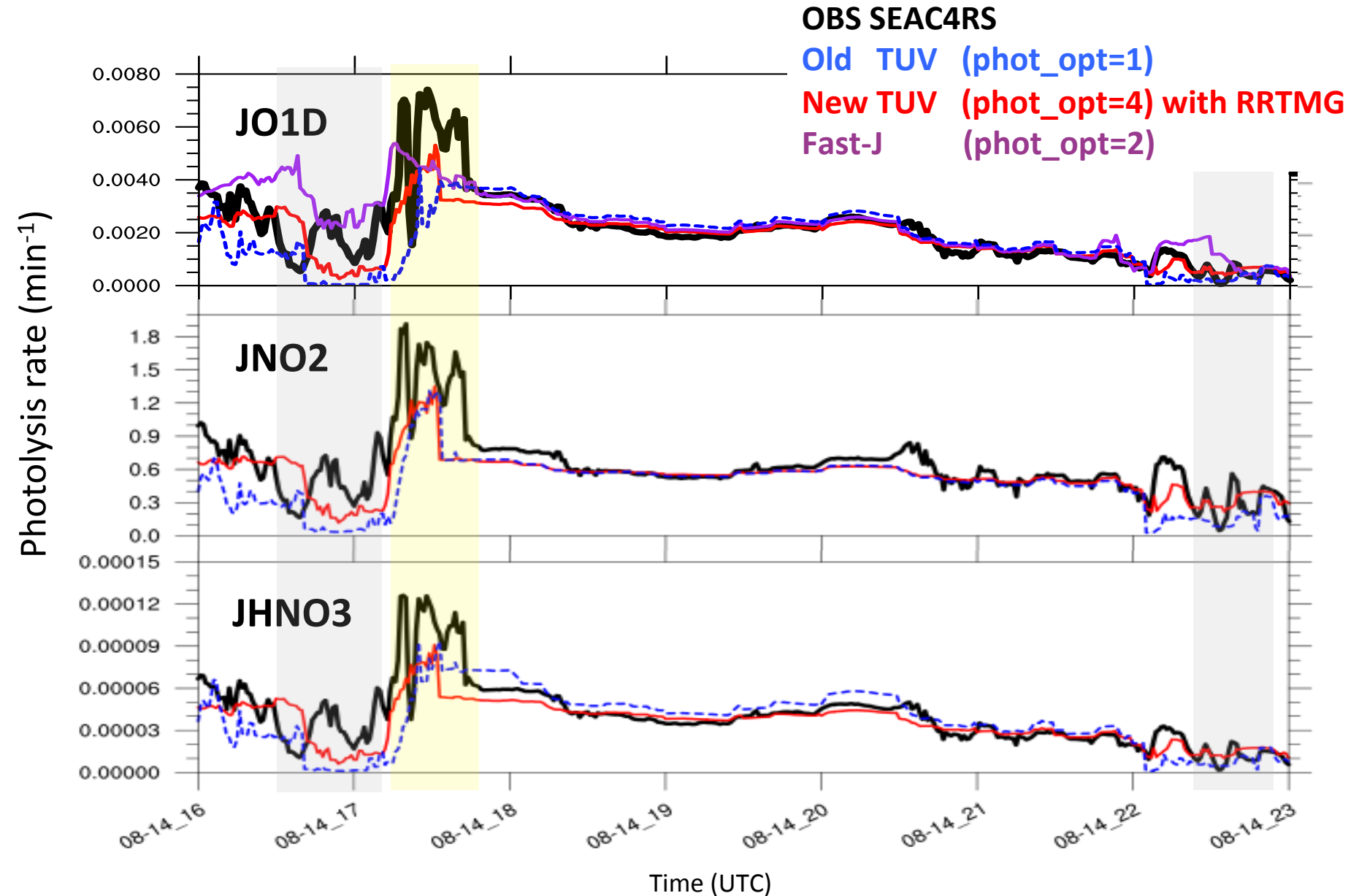
Comparison with SEAC4RS flight of 14 Aug. 2013



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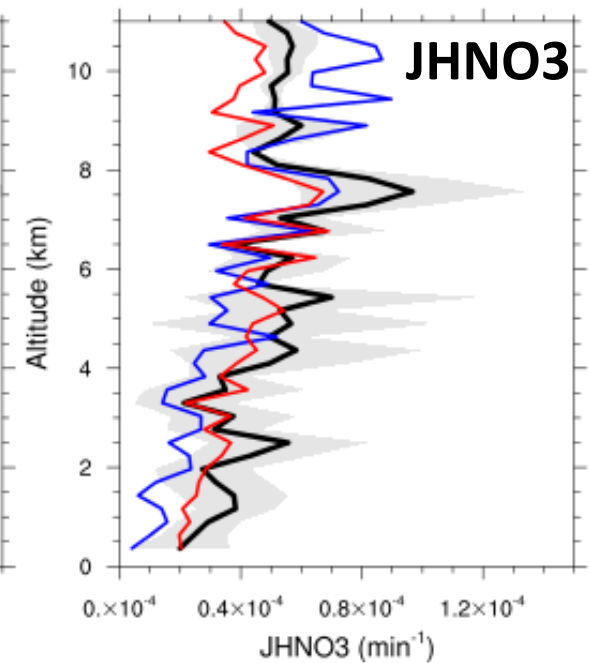
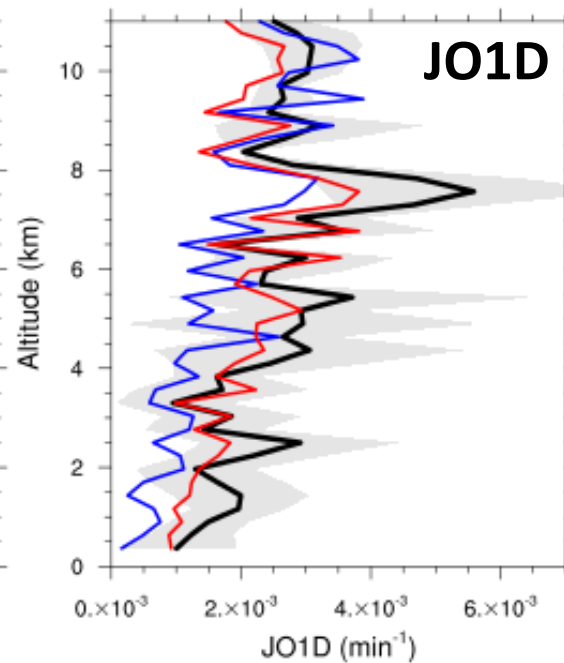
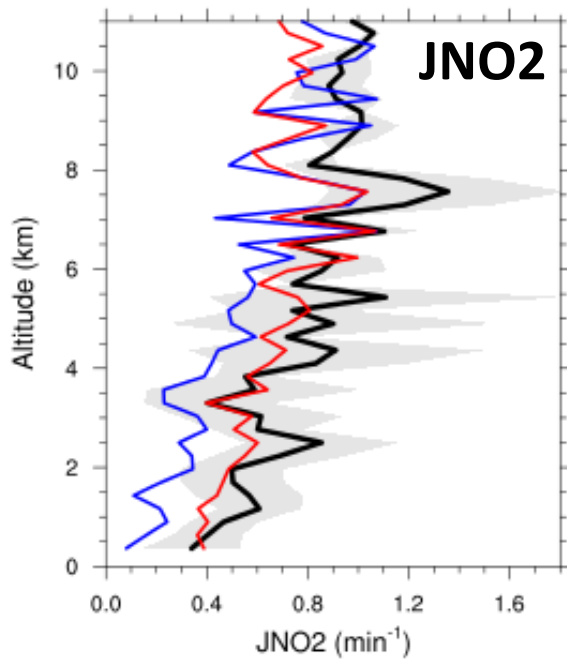
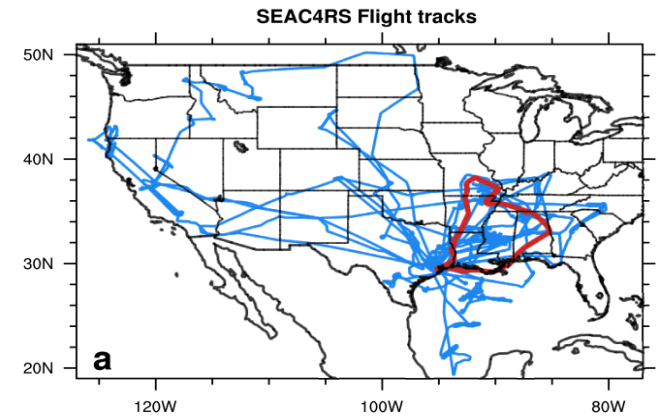


Comparison for all SEAC4RS flights

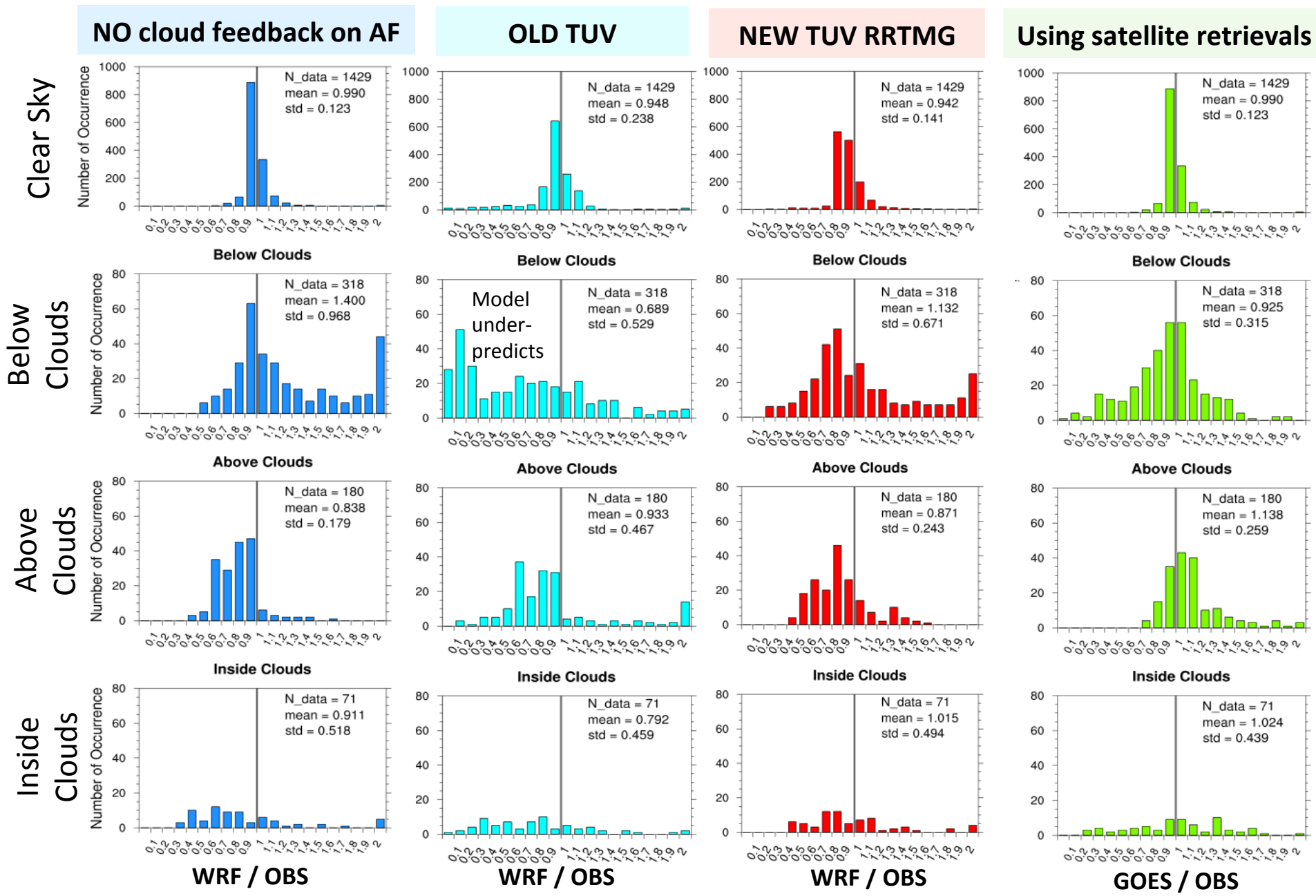
OBS SEAC4RS

Old TUV (phot_opt=1)

New TUV (phot_opt=4) with RRTMG

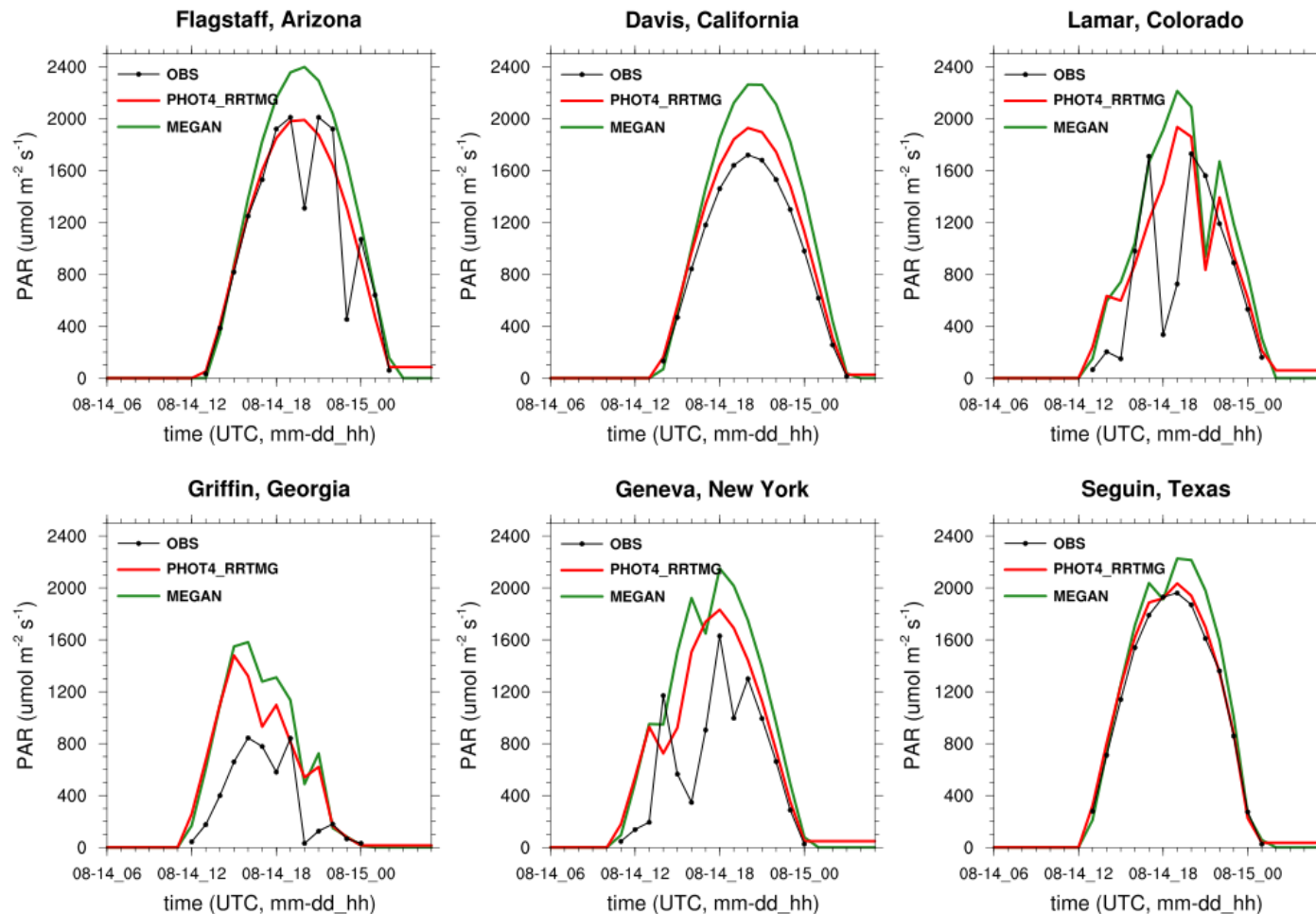


Ratio of WRF / OBS Actinic Flux for all SEAC4RS flights

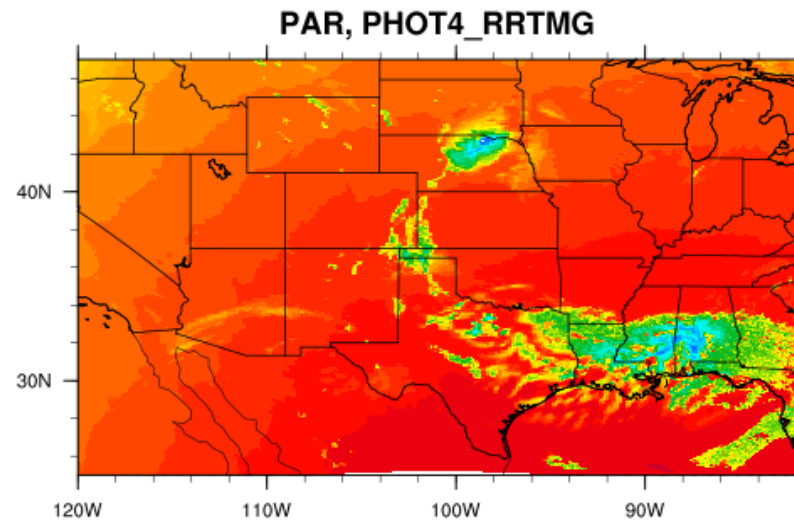
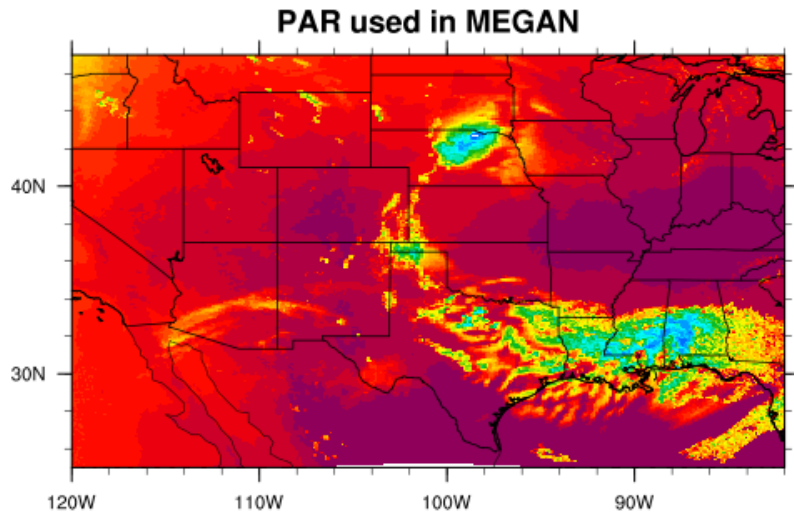


Comparison of PAR (Photosynthetically active radiation)

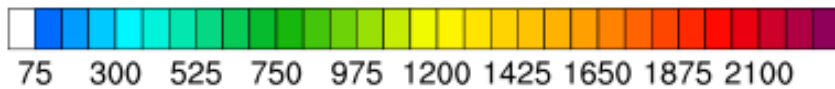
- Observed PAR (USDA UV-B Monitoring Network over the US)
 - WRF: used in MEGAN $PAR = 4.766 * 0.5 * SWDOWN$
 - WRF: calculated from new TUV RRTMG phot_opt=4
- 14 August 2013



PAR radiation – 14 August 2013



PAR ($\mu\text{mol m}^{-2} \text{s}^{-1}$)



10-20% difference in simulated PAR between values used in the MEGAN model for biogenic VOC emissions, and those predicted by TUV.
=> Effect on BVOC emissions?

Conclusions and next steps

Updated TUV (phot_opt = 4):

- Offers a more flexible framework for updating quantum yields and cross sections and adding new photolysis reactions through KPP
- Provides computationally efficient treatment of the subgrid cloud effects
- Consistency between TUV radiation with RRTMG and MEGAN emissions
- Better agreement with observations in clear sky and cloudy conditions compared to old TUV (photolysis are less attenuated below clouds)
- Updates will be made available in the next major model release

Next Steps

- Evaluate the effect on ozone production caused by changes in both photolysis and PAR values.
- Seek improvements in photochemistry through data assimilation of actinic fluxes or better cloud fields.

Results of the offline TUV driven by GOES clouds

